Effects of stoichiometric doping in superconducting Bi-O-S compounds

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Abstract. Newly discovered Bi-O-S compounds remain an enigma in attempts to understand their electronic properties. A recent study of Bi₄O₄S₃ has shown it to be a mixture of two phases, Bi₂OS₂ and Bi₃O₂S₃, the latter being superconducting [W. A. Phelan et al., J. Am. Chem. Soc. 135, 5372 (2013)]. Using density functional theory, we explore the electronic structure of both the phases and the effect of the introduction of stacking faults. Our results demonstrate that the S₂ layers dope the bismuth-sulphur bands. The bands at the Fermi level are of clear two-dimensional character. One band manifold is confined to the two adjacent, square-lattice bismuth-sulphur planes,

a second manifold is confined to the square lattice of sulphur dimers. We show that the introduction of defects in the stacking does not influence the electronic structure. Finally, we also show that spin-orbit coupling does not have any significant effect on the states close to the Fermi level at the energy scale considered.

1. Introduction

The recent discovery of superconductivity in two new compounds, $Bi_4O_4S_3$ [1, 2, 3, 4] and $LaO_xF_{1-x}BiS_2$ [5, 6, 7, 8], has raised great interest. All those compounds share the same structure: a stacking of alternating BiS_2 bilayers and spacer layers. Superconductivity is thought to arise from doping of the BiS_2 bilayers. Electron doping in $Sr_{1-x}La_xFBiS_2$ [9], and $La_{1-x}M_xOBiS_2$ (M = Ti, Zr, Hf, Th) [10] results in superconductivity. The versatility of the $La(O,F)BiS_2$ system has been demonstrated by replacement of La with a range of lanthanide Ln^{3+} ions [11, 12, 13, 14, 15]— a maximum T_c of 10.6 K is reported in $LaO_xF_{1-x}BiS_2$ for x = 0.5.

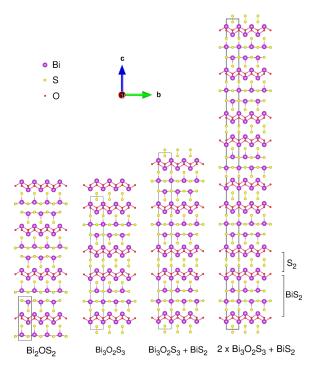


Figure 1. Crystal structure of Bi_2OS_2 , $Bi_3O_2S_3$, $Bi_3O_2S_3$ with one stacking fault per unit cell of $Bi_3O_2S_3$, and $Bi_3O_2S_3$ with one stacking fault per two unit cell of $Bi_3O_2S_3$, represented with Vesta [16]. They are formed of a stacking of BiS_2 , S_2 and Bi_2O_2 layers, the latter being a spacer layer. These structures implement four different S_2/BiS_2 ratios : 0, 1, 2/3, 4/5, i.e. four different frequencies of occurrence of stacking faults. The black squares represent the unit cells used in the calculations.

Electronic structure calculations have mainly focused on the $\text{LaO}_x\text{F}_{1-x}\text{BiS}_2$ (x = 0, 0.5) materials partially due to uncertainties in the composition of the Bi-O-S superconducting phase [17, 18]. These calculations indicated that the superconducting electrons are a mixture of Bismuth $6p_{x,y}$ and Sulphur $3p_{x,y}$ states [17, 18]. These form 8 bands, four of which sit under the Fermi level, when the other four either are above the Fermi level or cross it. The electronic and thermodynamic properties of this group of materials are indeed exciting and enigmatic and suggestions range from spin-fluctuation

mediated superconductivity [19] to proximity to ferroelectricity and charge density wave (CDW) instabilities [20, 21].

The coupling mechanism has been investigated in various ways. Electron-phonon interactions have been calculated in La(O,F)BiS₂, and yield a large electron-phonon coupling constant, suggesting superconductivity in this compound is strongly coupled and conventional [21, 20, 22]. Renormalisation-group calculations suggested triplet pairing and weak topological superconductivity [23, 24], a possibility studied in the context of quasiparticle interference [25]. Random phase approximation was applied to a two-orbital model [18], leading to an extended s-wave or d-wave pairing [19, 26].

The first superconductor of this family to be discovered, $Bi_4O_4S_3$, was studied in great detail [27, 28, 29, 30, 31, 32, 33]. Nevertheless, there were still doubts concerning its structure. However Phelan et al. recently published an extensive study of the chemistry of these compounds, and concluded that $Bi_4O_4S_3$ was actually a two-phased material [34]. The two phases are Bi_2OS_2 and $Bi_3O_2S_3$, the latter being assigned as the superconducting one. Single crystals of the superconducting phase have recently been synthesised [35].

Composed of BiS_2 , S_2 and Bi_2O_2 layers, the latter being a spacer layer (Figure 1), $Bi_3O_2S_3$ is made of alternating BiS_2 bilayers and S_2 layers, all separated by Bi_2O_2 spacer layers, whereas in Bi_2OS_2 only the two BiS_2 bilayers are present. Here, we call BiS plane a two-dimensional squared lattice containing bismuth and sulphur atoms. We call BiS_2 bilayer a structure containing two BiS planes and extra sulphurs localised on top or under each bismuth atom of the BiS planes. Figure 2 illustrates the difference between BiS plane and BiS_2 bilayer.

Experimental results [34] suggest that superconductivity is suppressed by the inclusion of additional BiS₂ bilayers disrupting the alternation of S₂ and BiS₂ sheets, which we call stacking faults. Phelan et al. studied these stacking faults in details, in particular displaying TEM data that shows without ambiguity the variations in stacking. Stacking faults play a significant role in superconducting properties of several families of materials [36]. For example, changes to the ground state have been observed in the presence of stacking faults in other systems, including modifications to superconductivity [37, 38].

In this paper we explore the electronic structure of the recently experimentally determined superconducting and non-superconducting phases of the Bi-O-S compounds. The role of introducing additional BiS₂ bilayers in the superconductor Bi₃O₂S₃ is also investigated, as suggested by Phelan et al. in their experimental work [34]. Like in the La(O,F)BiS₂ systems we find that the BiS₂ bilayers play a critical role in conduction. Our analysis indicates that the S₂ sheets give rise to electron carriers in the bismuth-sulphur bands. These electrons are localised within the BiS planes rather that through the BiS₂ bilayer. Interruption of the BiS₂, S₂ stacking sequence is shown not to influence the electronic structure.

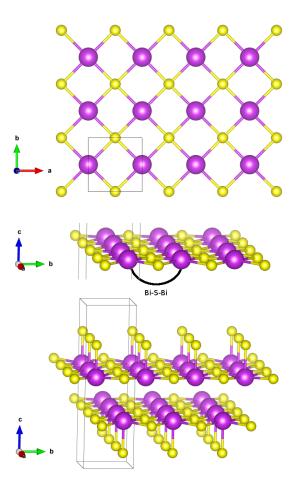


Figure 2. BiS plane seen from above (top), in perspective (middle), and BiS₂ bilayer seen in perspective (bottom) represented with Vesta [16]. One BiS₂ bilayer contains two BiS planes, and one extra S atom on top (or below) each Bi atom. The Bi-S-Bi angle is highlighted in the middle plot.

2. Methods

Band structures and density of states for Bi₂OS₂ and Bi₃O₂S₃ were calculated, using the SIESTA method [39, 40], implementing the generalized gradient approximation (GGA) in the shape of the Perdew, Burke, and Ernzerhof (PBE) functional. It uses norm-conserving pseudopotentials to replace the core electrons, while the valence electrons are described using atomic-like orbitals as basis states at the double zeta polarized level. In the case of the two known structures, experimentally determined structural parameters were used in our calculations [34]‡ Spin polarized calculations were found to

‡ For the first phase, Bi₂OS₂, the space group is P4/nmm, and the lattice parameters are a=3.9661 Å and c=13.798 Å . The atoms coordinates are Bi1 : (0.25,0.25,0.59122), Bi2 : (0.25,0.25,0.12989), O : (0.75,0.25,0.5), S1 : (0.25,0.25,0.328), S2 : (0.25,0.25,0.9063). For the second phase, Bi₃O₂S₃, the space group is I4/mmm, and the lattice parameters are a=3.96721 Å and c=41.2847 Å. The atoms coordinates are Bi1 : (0,0,0.05699), Bi2: (0,0,0.20865), Bi3 : (0,0,0.38245), O : (0,0.5,0.0852), S1 : (0,0,0.14595), S2 : (0,0,0.2870), S3 : (0,0,0.47748). All the occupancies are equal to one.

be unnecessary. We used a k-point grid of 16x16x5 for Bi_2OS_2 , of 16x16x2 for $Bi_3O_2S_3$ and of 16x16x1 for the two other structures, after showing the electronic structure was well converged using them. We used an electronic temperature of 7 K and a real space mesh cutoff of 300 Ry for the real-space integration, necessary to calculate the matrix elements for some terms of the Hamiltonian. The convergence of these parameters was also checked.

In order to estimate the influence of spin-orbit coupling, we also performed all electron calculations on $\mathrm{Bi_3O_2S_3}$ using full-potential linearised augmented plane waves, including spin-orbit coupling through a second-variational scheme, as implemented in the Elk program [41]. We used the PBE functional and the same k-point grid we used with SIESTA, along with a carefully converged high-quality set of parameters.

3. Crystal structure

The structures of the key materials for this study, Bi₂OS₂ and Bi₃O₂S₃, are taken from the experiments in [34]. In order to simulate how stacking faults perturb the electronic structure, we also investigated materials where the relative number of BiS₂ bilayers and S₂ layers was changed, as suggested by experiment, in particular by TEM data [34]. We did not use relaxed structures in order to keep consistency with the structures directly measured with X-rays.

In $Bi_3O_2S_3$, the stacking is 2 x BiS_2 / Bi_2O_2 / S_2 / Bi_2O_2 / 2 x BiS_2 / Bi_2O_2 / S_2 / Bi_2O_2 (Figure 1). The sequence is repeated twice because of symmetry: S_2 layers shift the Bi_2O_2 layers next to them on the x and y axis. Therefore the unit cell has to be doubled so that the stacking of one cell on another is possible.

To modify the ratio of S_2 layers over BiS_2 bilayers, we added a BiS_2 bilayer, with its corresponding Bi_2O_2 spacer layer, at the top of $Bi_3O_2S_3$. We therefore obtain the following stacking: $2 \times BiS_2 / Bi_2O_2 / S_2 / Bi_2O_2 / 2 \times BiS_2 / Bi_2O_2 / S_2 / S_2 / Bi_2O_2 / S_2 / Bi_2O_2 / S_2 / S_2 / Bi_2O_2 / S_2 /$

The structure of BiS₂ bilayers is different in Bi₂OS₂ and Bi₃O₂S₃, the most noticeable difference being the Bi-S-Bi angle in the BiS planes (Figure 2), which is larger in the parent phase. We thus have had to choose which BiS₂ bilayer to add in the compounds with a modified stacking: Bi₃O₂S₃+BiS₂ and $2xBi_3O_2S_3+BiS_2$ (Figure 1). We did calculations for both, and the results were extremely similar (e.g. the bands are less than 0.1 eV apart). The results discussed below are the ones for Bi₃O₂S₃'s BiS₂ bilayers, as the stacking faults appear in this compound.

4. Electronic structure

The band structures calculated with Elk and SIESTA are in very close agreement (Figure 3, right hand side). The influence of spin-orbit coupling is very limited, with a difference under 0.15 eV between energy levels with and without spin-orbit coupling (Figure 3, left hand side). The presence or absence of spin-orbit coupling has no impact on any of the features of the electronic structure discussed below. In the following we concentrate on electronic structures calculated with SIESTA.

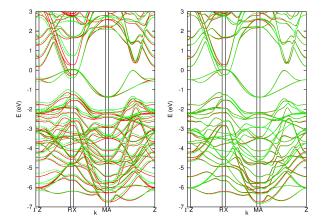


Figure 3. Left: band structure of $Bi_3O_2S_3$ calculated with Elk, with (red) and without (green) spin-orbit coupling. Right: band structure of $Bi_3O_2S_3$ calculated with Elk (red) and with SIESTA (green) without spin-orbit coupling. In all band structures presented in this paper, the Fermi level is at zero energy.

4.1. Band structures

Bi₂OS₂ and Bi₃O₂S₃ only differ by the replacement in the second compound of one out of two BiS₂ bilayer by an S₂ layer. In terms of stoichiometry, as noted by Phelan et al. [34], Bi₂OS₂ can be written BiOBiS₂, just as the parent phase of the lanthanum compound, LaOBiS₂.

The band structure of Bi_2OS_2 (Figure 4) has features close to the Fermi energy which are similar to those of the undoped $LaOBiS_2$ phase [17, 18]. For all band structures calculated, of specific interest is the set of bands crossing or just above the Fermi level near the R and X points, and approaching the Fermi level between the A and Z points (Figure 4). These were also found in $LaO_xF_{1-x}BiS_2$ [17, 18], and were shown in these compounds to be composed of bismuth $6p_{x,y}$ and sulphur $3p_{x,y}$ orbitals. Partial density of states calculations enabled us to confirm that in all the compounds considered in this study these bands are composed of bismuth $6p_{x,y}$ and sulphur $3p_{x,y}$ orbitals, confirming the assignment done in [18] (Figure 4).

In the first case, Bi₂OS₂, these BiS bands are just above the Fermi level, at around 0.3 eV at their minimum. This is very close to the corresponding minimum in LaOBiS₂

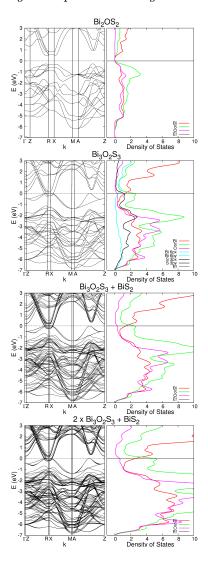


Figure 4. Band structures and densities of states projected onto the basis orbitals for Bi (red), S (green) and O (lilac) for the four simulated compounds. From top to bottom, they correspond to the structures from left to right in Figure 1. The two last compounds are simulations of stacking faults for two different frequencies in Bi₃O₂S₃. In Bi₃O₂S₃ without stacking faults, we plot the partial density of states for the bismuth $6p_{x,y}$ (light blue) and sulphur $3p_{x,y}$ orbitals (black). In each case, the p_x and p_y orbitals are indistinguishable. At the Fermi level, they each contribute approximately half of the density of state of the corresponding specie. These projections have been omitted in the other plots for clarity. The y-axis has been chosen so that the Fermi level is at zero energy.

(0.2 eV) [17, 18]. The gap is 0.7 eV wide, twice as large as in the LaOBiS₂ case (0.37 eV) [17, 18].

In $Bi_3O_2S_3$, one of these BiS bands crosses the Fermi level near the R and X points, just as in $LaO_{0.5}F_{0.5}BiS_2$ [17, 18]. At these points it interacts with two other bands. These are in great majority composed of S 3p states, as the projected density of states shows ("projected" refers to decomposing the total density of states in contributions

from the different basis orbitals of the different atoms). These bands are not present in the parent phase, Bi_2OS_2 ; they come from the S_2 layer. These two bands therefore correspond to the partly filled π antibonding states in the S_2 dimers. If fully occupied, this would formally correspond to S_2^{2-} dimers.

4.2. Local densities of states

To confirm these bands are coming from the S_2 layer, we plotted the local density of states ("local" refers to decomposing the total density of states into real space), integrated in the range of energies between -1.5 eV and 0 eV (Figure 5). The chosen range of energies contains the bands we are interested in.

The results are quite clear: it is indeed the S_2 layer that gathers all the electron density. Moreover, this density is organised as a torus oriented along the z axis around each S atom (the fact that the density decreases near the atom comes from the pseudopotentials), which shows that the orbitals lying at or just above the Fermi level correspond mainly to p_x, p_y electrons on the S_2 dimers (oriented along z). The concentration of the electron density on the atoms rather than on the S–S bond could be indicative of a weakening of the S–S interaction.

We also plotted the local density of states in real space for the range of energies between 0 eV and 1.5 eV (Figure 5). This range contains the characteristic bands present in all the BiS_2 -based compounds. We can clearly see that the charges are localised in the BiS_2 bilayer, more precisely in the two BiS planes. The S atoms in the BiS_2 bilayer which are outside the BiS planes gather as few electrons in that range of energy as the S atoms in the S_2 layer.

4.3. Pinning of the Fermi level

In the band structures of $Bi_3O_2S_3$ with and without stacking faults, we observe that the top S_2 band touches the Fermi level, without crossing it, approximately at equal distance from A and Z. Interestingly, this band is perfectly flat along the z direction: this S_2 related band has very marked two-dimensional character, and therefore it shows the characteristic discontinuous 2D Van Hove singularity (very clear at the Fermi level in the three lower panels of Fig. 4), which seems to be pinning the Fermi level. Such a discontinuity is suggestive of possible instabilities along the (1,1,0) direction.

4.4. Influence of stacking faults

The electronic structure of $Bi_3O_2S_3$ with an added stacking fault is very close to that of $Bi_3O_2S_3$ (Figure 4). The only difference is that some bands below the S_2 bands are raised slightly. It has strictly no impact on the bands close to the Fermi level.

5. Conclusions

The main conclusions of this study are:

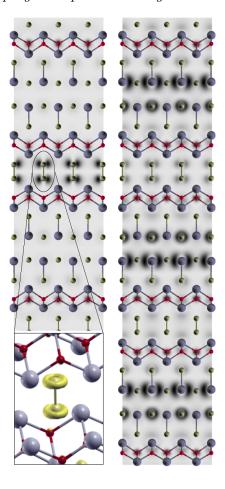


Figure 5. Local density of states of $Bi_3O_2S_3$ in real space integrated in the energy ranges [-1.5,0] (a) and [0,1.5] (b), plotted with XCrySDen [42]. Framed is a 3D plot of an isosurface centred on the S-S dimer. In the first case, the S_2 layer gathers most of the density. In the second case it is the BiS planes, without the extra S atoms, that gather the density. Half of the atoms represented are not in the plane corresponding to the density plot.

(i) The key result of this study is that, in spite of it being a nominally stoichiometric composition (no doping), the S_2 layers in the $Bi_3O_2S_3$ compound act as effective dopants of the parent insulating compound. They push a p_x, p_y -like band of the S_2 layer onto the bottom of the conduction band, thereby transferring part of the electron density from the S_2 dimers to the BiS planes. This doping is in agreement with experiments, that show that $Bi_3O_2S_3$ is metallic [34].

Furthermore, the experimental observation of an anomalously short S–S distance in the S_2 layers is also consistent with electron depletion of the S_2 layer. Indeed, a substantial residual force is observed for the S atoms in the S_2 layer in the direction of elongating the distance, which suggests that the real system has a larger depletion of electrons from the S_2 bands than the one obtained in the calculations.

(ii) The electronic structure close to the Fermi level is not modified by stacking faults. This indicates that the changes in the superconducting state caused by these are not

related to the electronic structure

(iii) Finally, the calculations of density of states in real space indicate that the BiS bands, shown to be central for superconductivity in these compounds [18], actually come from the BiS planes. Therefore finding other structures having such planes would be very interesting in order to tune their interactions differently, and maybe transform their behaviour.

In summary, we studied in detail the electronic structure of the newly discovered superconductor $Bi_3O_2S_3$. We find the bismuth-sulphur bands corresponding to the BiS planes are doped by the S_2 dimers, unlike the other superconducting members of the BiS_2 family which are non-stoichiometric. We studied the influence of stacking faults on this material and found it does not influence the electronic structure. Finally, the fact that the electrons responsible for superconductivity are localised in the 2D bands of BiS planes suggests the possibility of other superconductors with BiS structural units.

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