

Gas phase Raman spectroscopy : comparison of continuous wave and cavity based methods

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Abstract: Comparison of cavity-enhanced Raman spectroscopy to continuous wave detection for gas phase molecules in air. We show continuous measurements with calculated emission and discuss the potential benefits (two orders more signal) of using a cavity. © 2018 The Author(s)

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Spontaneous Raman scattering is a valuable diagnostic tool that has found use in many different fields of research and industry. In chemical kinetics, vibrational spectroscopy has found use for Raman flame measurements in combustion and Thomson scattering measurements of low-density plasmas. In addition to this, the chemical pathway for many reacting flows has much interest for simulations, design and automation. Raman allows the ability to follow simple gas species such as oxygen, nitrogen and carbon dioxide during process control providing a powerful method for quantitative gas phase analysis and trace gas detection for environmental monitoring. Safety monitoring of hydrogen gas levels, isotope tracer studies and monitoring radioactive tritium is paramount not only to atmospheric sensing research however to the health and safety of many work places in industry. Surface chemistry and catalysis include refining the theoretical analysis of molecular energy levels and potential energy hypersurfaces. Furthermore, due to its high-resolution, there is potential for increased fundamental scientific understanding of vibrational interactions and intramolecular vibrational energy redistribution on faster timescales. These different fields tend to use high power laser (CW) systems with large footprints. Here we propose a technique called Cavity Enhanced Raman Spectroscopy (CERS). Including the already increasing range of applications, this technique would also motivate real time, low cost, small footprint in situ measurements in a broad range of other scientific and industrial fields from NEMs/MEMs systems, organic electronics and Nano/Microscale chemistry to planetary atmospheric detection, exo-meteorology, deep-sea explorations [1], human breath analysis [2], turbine power plant [3] and fermentation gases [4]. Real time analysis capabilities, and continuous Raman signals with low power diodes, makes the cavity technique none invasive and applicable for future applications.

A detailed calculation of the rate of emission of Raman photons in such systems can be found in [5]. The rate of emission per molecule in free space scaled by the input power of the laser is given by

$$\frac{\gamma_s}{P_{in}N} = R_f \frac{\lambda_s^4}{\Delta\lambda_s^2} \frac{d\sigma}{d\Omega}, \quad (1)$$

where $R_f = 3d_0^3/64hc^3$ is a constant based on the optical arrangement, $d\sigma/d\Omega$ is the differential Raman scattering cross section, with λ_s and $\Delta\lambda_s$ being the Stokes wavelength and linewidth, respectively. For a typical setup one can expect the magnitude of R_f to be $\approx 5 \times 10^{16}$. The maximum amount of photons are expected for a long wavelength laser with high power over a narrow linewidth. Note $d\sigma/d\Omega$ also has a wavelength dependence. There are three main factors that contribute to the enhancement of spontaneous Raman scattering within a cavity: (1) Resonant recirculation describes the power build up within the cavity, (2) Purcell effect describes the increase in the rate of spontaneous emission due to the presence of a cavity, (3) Double resonance condition describes that the cavity length must be set such that the number of free spectral ranges equals the shift of the Raman mode of interest. The rate of emission per molecule in a cavity scaled by the input power of the laser is given by

$$\frac{\gamma_s}{P_{in}N} = R_c \frac{\lambda_s^6}{\Delta\lambda_p^{(c)} \Delta\lambda_s} \frac{d\sigma}{d\Omega}, \quad (2)$$

